NEW TYPES OF SESQUITERPENE LACTONES AND OTHER CONSTITUENTS FROM TRICHOGONIA SPECIES*

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Abstract—The investigation of five *Trichogonia* species afforded seven new furanoheliangolides, three dilactones of a new type, four pairs of epimeric rearranged germacranolides, a new sesquiterpene acid and nine new phydroxyacetophenone derivatives, two of them of a new type with a furan ring. The structures were elucidated by spectroscopic methods and some chemical transformations. The structure of one of the dilactones was established by X-ray analysis. The biogenetic pathways and the chemotaxonomic importance of the new compounds are discussed.

INTRODUCTION

The mainly Brazilian genus Trichogonia (DC) Gardn. [1], earlier a section of Kuhnia [2], is placed in the Gyptis group (tribe Eupatorieae) [3]. No chemical investigations have been reported previously on these plants. In continuation of our chemosystematic studies of the tribe Eupatorieae, we have studied the constituents of five Trichogonia species to determine if the chemistry may give indications of the relationships in this diverse group.

RESULTS AND DISCUSSION

The aerial parts of T. prancii afforded a very complex mixture of sesquiterpene lactones, which could be only partly separated. The main constituents were the lactones 1a and 4, which could not be separated even by HPLC. The ¹H NMR spectrum of **1a** (Table 1) showed that a heliangolide with a furanone ring was present, which was further supported by the characteristic UV maximum and the IR bands. The nature of the ester residue also followed from the typical ¹H NMR signals, while the stereochemistry at C-6 through C-9 was deduced from the observed couplings. The relative position of the ester residue followed from the chemical shift of H-8, which was assigned by spin decoupling in the usual way. The presence of a 6,12-lactone was indicated by the similarity of the chemical shifts of H-5 through H-8 with those of known lactones of this type [4–6]. The stereochemistry at C-8 and C-9 was supported by the presence of a hydrogen bond between the 9-hydroxy and the 1-keto group, which followed from the IR band and the large H-9,OHcoupling.

Acetylation of 1a and 4 using 4-pyrrolidinopyridine as a catalyst [7] afforded the acetates 2 and 5, which were identical with two other lactones also present in the extract. Again a separation of these two lactones was not possible. The ¹H NMR data of 4 (Table 1) clearly showed that the 11,13-double bond in la was epoxidized. All couplings were nearly the same as in the spectrum of 1a. However, the orientation of the epoxide ring could not be assigned with certainty. Most probably a β -epoxide was present as H-6 seemed to be deshielded, if the chemical shifts were compared in the spectra of 1a and 4, while the signals of H-8 and H-9 were, as expected, at slightly higher fields in the spectrum of 4. A further heliangolide was an isomer of 1a, in which the relative positions of the methacrylate and the hydroxy group were interchanged. The ¹H NMR data (Table 1) and those of the acetate obtained by acetylation showed that the structure was 3a. The free C-8 hydroxy group caused a considerable downfield shift of the H-6 signal, again supporting the proposed stereochemistry at these centres. Compounds 1a, 2 and 3a were derivatives of atripliciolide, which is the 8-desacyl-9-desoxy derivative of 1a [6].

The structure elucidation of three further sesquiterpene lactones caused more difficulties. ¹H NMR spectroscopic investigations (Table 2) finally led to the structures 6, 7 and 8a. Mild acetylation of 6 afforded 7, indicating identical stereochemistry at all centres in both compounds. Spin decoupling allowed the assignment of the signals of H-5 through H-9 in the usual way starting with irradiation of the H-7 signal. The chemical shifts of H-5 and H-6 obviously required an acceptor group at C-3, while a quartet at δ 5.01 and a doublet at 1.39 (3 H) could be explained, if a seco-germacranolide with a tenmembered lactone ring were present. Inspection of a model showed that H-6 should be strongly deshielded by the lactone oxygen, the C-10 hydroxy and the C-8 ester group. The observed couplings were in good agreement with the angles deduced from the model. We have given the name trichogoniolide to compound 6.

^{*}Part 332 in the series "Naturally Occurring Terpene Derivatives". For Part 331 see Bohlmann, F., Jakupovic, J., Dhar, A. K., King, R. M. and Robinson, H. (1981) *Phytochemistry* 20, 1081.

Table 1. ¹H NMR spectral data of compounds of 1a, 1b, 2, 3a, 3b, 4, 5 and 41 (400 MHz, TMS as internal standard, CDCl₃)

	1a	1b	2	3a	3b	4	5	41
H-2	5.64 s	5.63 s	5.64 s	5.61 s	5.67 s	5.65 s	5.63 s	5.60 br.s
H-5	5.97 dq	5.98 dq	6.00 dq	6.00 dq	6.02 dq	6.03 dq	6.06 dq	{2.58 ddd 2.06 br.d
H-6	4.95 ddq	5.03 ddq	5.24 ddq	5.73 ddq	5.36 ddq	5.03 ddq	5.21 ddq	4.27 ddd
H-7		3.65 <i>dddd</i>		3.46 <i>dddd</i>			3.26 dd	3.30 m
H-8	5.10 dd	5.18 dd	5.38 dd	4.16 dd	5.50 dd	4.98 dd	5.13 dd	5.30 br.d
H-9	4.19 dd	4.8 dd	5.33 d	5.28 d	5.30 d	3.98 dd	5.09 d	4.18 dd
H-13	6.36 d	6.39 d	6.40 d	6.37 d	6.44 d	3.47 d	3.42 d	6.40 d
H-13'	5.72 d	5.72 d	5.83 d	5.70 d	5.89 d	3.41 d	3.36 d	5.77 d
H-14	1.67 s	1.68 s	1.53 s	1.48 s	1.53 s	1.65 s	1.51 s	1.66 s
H-15	2.06 dd	2.07 dd	1.07 dd	2.05 dd	2.09 dd	2.09 dd	2.07 dd	1.40 d
OCOR	6.03 dq	6.17 qq	6.07 dq	6.45 dq	6.28 dq	6.12 dq	6.14 dq	6.16 qq
	5.68 dq	1.94 dq	5.63 dq	5.74 dq	5.72 dq	5.72 dq	5.73 dq	1.95 dq
	1.88 dd	1.85 dq	1.89 dd	2.02 dd	1.98 dd	1.89 dd	1.90 dd	1.85 dq
OAc		_	2.13 s		2.01 s	-	2.14 s	
ОН	3.55 d	3.40 d				3.61 d		3.90 d

J (Hz): Compounds: 1/2: 5.6 = 4; 5.15 = 1.7; 6.7 = 3.5; 6.15 = 1.7; 7.8 = 2; 7.13 = 3; 7.13' = 2.7; 8.9 = 3; (1: 9.0H = 11.5); compounds 3a/3b: 5.6 = 3.5; 5.15 = 1.8; 6.7 = 3.5; 6.15 = 1.8; 7.13 = 3.5; 7.13' = 3.5; 7.13' = 3.5; compounds 7.13' = 3.5; 7.13' = 3.5; 7.13' = 3.5; 7.13' = 3.5; compounds 7.13' = 3.5; 7.13' =

RO
$$\frac{14}{10}$$
 OAc RO $\frac{14}{10}$ O Meacr O $\frac{3}{10}$ O $\frac{6}{10}$ O $\frac{7}{10}$ O $\frac{3}{10}$ O $\frac{4}{10}$ O $\frac{1}{10}$ O $\frac{3}{10}$ O $\frac{4}{10}$ O

R = Ac

The ¹H NMR data of 8a (Table 2) showed that this lactone must be an isomer of 6. The H-1 signal was a double quartet, its position already indicated a proton under a free hydroxy group. Irradiation of the doublet at δ 2.49 collapsed the signal at 3.98 to a quartet. The latter was coupled with the methyl doublet at 1.33. The presence of a nine-membered lactone was supported by the observed downfield shift of the 10-methyl signal. Acetylation afforded the diacetate 8b, its ¹H NMR data further supported the proposed structure of 8a, which we have named isotrichogoniolide-9-O-acetate. The stereochemistry at C-1 in 6, 7 and 8a could not be determined. Xray analysis of 7 (Fig. 1) showed that the 1-methyl was β orientated and therefore the same configuration was also very likely in 8a. Most probably these lactones were formed by degradation of 1a (Scheme 1). So far the only known seco-germacranolide is the diol pycnolide [8]. The aerial parts also contained germacrene D, bicyclogermacrene, α-humulene, taraxasterol and taraxasteryl acetate. The roots afforded bicyclogermacrene, α-humulene, lupeyl acetate and two further compounds with molecular formulae C₁₄H₁₂O₃ and C₁₆H₁₆O₄. The ¹H NMR data (Table 3) showed that these compounds must be very similar. The former had an aldehyde group which was replaced in the second compound by a CH₂OAc group. The signals of the aromatic protons indicated the presence of unsymmetrically trisubstituted benzene derivatives, one substituent being a vinyl group, which must be far away from the aldehyde group, as no considerable shift differences in the ¹H NMR spectra of the two compounds were observed, while the narrowly split downfield signal at δ 8.10 in the spectrum of the aldehyde was at higher field in the spectrum of the acetate. In the latter, this signal was further split to a double triplet indicating an allylic coupling with the CH₂OAc protons. The signal at 8.10 was coupled with a broadened singlet at 7.30, which was strongly shifted after addition of Eu(fod)₃. All data agreed most accurately with the structures 9 and 10a. This

Table 2. ¹H NMR spectral data of compounds 6, 7, 8a and 8b (400 MHz, TMS as internal standard, CDCl₂)

	6	7	$7(C_6D_6)$	8a	8b
H-1	4.89 q	5.01 q	4.84 q	3.98 dq	5.19 q
H-2	1.51 d	1.39 d		1.33 d	$1.38 \dot{d}$
H-5	6.29 dq	6.33 dq	6.03 dq	6.39 dq	6.42 dq
Н-6	6.39 ddq	6.21 ddq	6.34 ddg	6.17 ddq	6.13 dda
H-7				3.20 dddd	
H-8				5.59 dd	
H-9				5.63 d	
H-13	6.25 d	6.24 d	6.19 d	6.33 d	6.32 d
H-13'	5.71 d	5.69 d	5.22 d	5.68 d	5.67 d
H-14	1.46 s	1.43 s	0.96 s	1.75 s	1.83 s
H-15	1.94 dd	1.97 dd	1.75 dd	1.93 dd	1.94 dd
OMeacr	6.15 dq	6.03 dq	6.24 dq	6.00 dq	5.96 dg
	5.66 dq	5.58 dq	5.19 dq	5.58 dq	5.56 dq
	1.92 dd	1.86 dd	1.88 dd	1.85 dd	1.83 br.s
OAc	_	2.14 s	1.66 s	2.16 s	2.08 s
					1.98 s
ОН		2.64 br.s		2.49 d	_

J (Hz): Compounds 6/7: 1,2 = 6.5; 5,6 = 4; 5,15 = 1.7; 6,7 = 1.5; 6,15 = 1.7; 7,8 = 3.5; 7,13 = 2; 7,13' = 1.7; 8,9 = 4; compounds 8a/8b: 1,2 = 6.5; 5,6 = 2.5; 5,15 = 1.8; 6,7 = 2.5; 6,15 = 1.8; 7,8 = 4; 7,13 = 2.5; 7,13' = 2; 8,9 = 5; OMeacr: $3_1',3_2'=3',4'\sim 1$.

assumption was confirmed by the catalytic hydrogenation of 10a, which led to the alcohol 12a, as could be deduced from the $^1\mathrm{H}$ NMR data (Table 3). However, only in a mixture of $\mathrm{CDCl_3-C_6D_6}$ were all the signals separated. Spin decoupling showed that the signal at δ 1.72 was

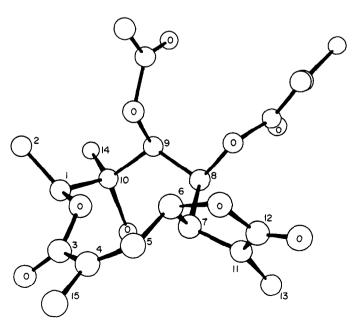


Fig. 1. Perspective drawing of compound 7.

Scheme 1.

coupled with two groups of signals, which must be assigned to CH₂OAc and CH₂OH. The proton which coupled with these groups was further coupled with a multiplet at 1.56. Irradiation of the latter collapsed the triplet at 2.62 to a singlet, clearly indicating the sequence H-9 through H-13. Acetylation of 12a afforded a symmetrical compound as could be seen from the ¹H NMR spectrum (Table 3). The formation of 12a obviously included a hydrogenolysis of the intermediate 11 (Scheme 2). Sodium boronate reduction of 9 gave the alcohol 10b, whose ¹H NMR data (Table 3) also supported the proposed structures. Compound 10a is probably formed in the plant starting with the prenylated p-

methoxyacetophenone 13 by oxidation and ring closure via 14 and 15 (Scheme 3).

The roots of T. grazielae afforded germacrene D and again several p-hydroxyacetophenone derivatives, only one being known, the carbinol 21 [9]. The structures of six further compounds (16–20 and 22) followed from the ^{1}H NMR data (Table 4). From the spectrum of 16 the presence of a dimethyl allyl side-chain was obvious. Furthermore, the typical signals of an angelate could be seen, which had to be placed in a benzylic position, following from the corresponding quartet at δ 5.91. The ^{1}H NMR data of 17 were similar to those of the known 4-acetyl chromanone [10]. However, the acetyl group was

Table 3. ¹H NMR spectral data of compounds 9, 10a, 10b, 12a and 12b (270 MHz, TMS as internal standard, CDCl₃)

	9	Δ*(9)	10a	Δ *(10a)	10b	12a*	12b
H-3	7.91 d	0.05	7.89 d	0.02	7.90 d	6.92 br.s	6.95 d
H-5	7.37 dd	0.02	7.31 dd	0.01	7.31 dd	6.93 br.d	7.00 dd
H-6	6.95 d	0.02	6.98 d	0.01	6.92 d	6.63 d	6.76 d
H-7	6.72 dd	0.03	6.71 dd	0.01	6.71 dd	2.52 q	2.57 q
H-8t	5.70 dd	0.03	5.69 dd	0.02	5.70 dd	${}$ 1.18 t	${}$ 1.21 t
H-8c	5.21 dd	0.03	5.18 dd	0.02	5.18 dd	f1.10 t	\\ \tag{1.21 \tau}
H-9		_	_	_		2.62 t	2.65 t
H-10	7.30 br.s	0.33	6.98 br.s	0.12	7.01 br.s	1.56 m	1.66 m
H-11			1 shares		- souther	1.72 ddda	lt 2.06 m
H-12 H-12'	8.10 d	}0.11	7.89 dt	0.12	7.46 dt	3.39 dd 3.34 dd	
H-13 H-13'	9.98 s	}0.35	5.02 br.s	0.40	4.62 br.s	4.15 dd 4.06 dd	4.10 dd
OAc	, common	MAP PRO	2.09 s	0.36		1.83 s	2.06 s
OMe	3.96 s	0.04	3.95 s	0.02	3.96 s	3.58 s	3.79 s

^{* 400} MHz, CDCl₃-C₆D₆(1:2).

J (Hz): Compounds **9,10a/b**: 3,5=2; 5,6=8.5; 7,8t=17; 7,8c=10; 8t,8c=1; $10,12=12,13\sim1$; compound **12a**: 5,6=8; 7,8=7.5; 9,10=8; $10,11\sim5$; 11,12=4.5; 11,12'=6; 12,12'=11; 11,13=4.5; 11,13'=6; 13,13'=11; **(12b**: 11,12=11,13=5; 11,12'=11,13'=6).

$$\begin{array}{c|c}
OMe & (H) \\
\hline
-H_2O \\
\hline
13 & 14
\end{array}$$

$$\begin{array}{c|c}
OMe \\
\hline
-H_2O \\
\hline
15 & HO
\end{array}$$

$$\begin{array}{c|c}
OMe \\
\hline
-H_2O \\
\hline
-H_2O \\
\hline
\end{array}$$

$$\begin{array}{c|c}
OMe \\
\hline
-H_2O \\
\hline
\end{array}$$

Scheme 3.

replaced by a vinyl group, as could be deduced from the typical signals. The substitution pattern clearly followed from the couplings and chemical shifts of the aromatic protons. The ¹H NMR data of 18a showed that we were dealing with an ortho-hydroxyketone (12.81 s). All other signals were in full agreement with the structure proposed. The ¹H NMR data of 19 and 20 were similar, except for the signals of the ester residue. While 19 was obviously an angelate, the situation in 20 only became clear in C₆D₆ at 400 MHz. Due to the chiral centre at C-7, the protons of the CH₂OAc group were not equivalent and therefore two complex signals were observed which were poorly separated. Irradiation at 1.83 collapsed the signals at 5.91 to a triplet and those at 5.27 and 5.22 to doublets. The structure of 22 followed directly from the ¹H NMR data. The free 7-hydroxy group caused a pronounced upfield shift of H-7 compared with those in 19 and 20. Compound 18a is closely related to 10a.

The aerial parts afforded germacrene D, α -humulene, aromadendrene, caryophyllene, herniarin, the *p*-hydroxyacetophenone derivative 13 and γ -muurolen-15-oic acid (23a), whose structure elucidation caused some

difficulties. From the ¹H NMR spectrum of the corresponding methyl ester 23b (Table 5), the presence of an iso-propyl group, a conjugated carboxyl group and a methylene group were deduced. The ¹³C NMR spectrum (see Experimental) showed no quarternary saturated carbon signals and four tertiary ones. Together with the ¹H NMR data this was an indication of the presence of a cadinane or a guaiane derivative. Careful spin decoupling in different solvents, also after addition of Eu(fod)₃, led to the partial structure A.

$$-CH_{A} = C(CO_{2}R)C - C - CH_{D} - C - A$$

$$+ H_{B'} + H_{C'} + H_{E}$$

The allylic position of H_B followed from the allylic coupling with H_A , while the presence of a neighbouring CH_2 group was deduced from the two observed vicinal couplings of $H_{B'}$. However, $H_{B'}$ only gave a broadened double doublet. Irradiation of the H_D signal showed in

Table 4.	¹ H NMR spectral data* of compounds 16–20 and 22 (270 MHz, 1 MS as int.	
	stand., CDCl ₃)	

	16	17	18a	18b	19	20†	22
H-3	7.16 d	7.88 d	7.76 d	7.61 d	7.78 d	7.82 d	7.84 d
H-5	7.18 dd	7.57 dd	7.58 dd	7.46 dd	7.50 dd	7.29 dd	7.55 dd
H-6	6.80 d	6.91 d	6.97 d	6.91 d	6.98 d	7.07 d	6.93 d
H-7	5.91 q	6.66 dd	6.68 dd	6.67 dd	5.94 q	5.97 q	4.89 q
H-8	$\left.\right\}$ 1.55 d	5.68 d	5.63 d	5.66 d	$\left.\right\}$ 1.58 d	${1.42 d}$	1.49 d
H-8′	J 1.55 	5.21 d	5.19 d	5.18 d)))	
H-9	3.30 br.d						
H-10	5.30 br.t	2.74 s	6.83 qq	6.62 qq	6.79 qq	6.68 qq	2.73 s
H-12	1.74 br.s	1. 47	2.07 d	1.98 d	2.07 d	1.64 d	1.46 s
H-13	1.74 <i>br.s</i> 1.70 <i>br.s</i>	}1.4/ s	2.22 d	2.24 d	2.21 d	2.09 d	1.40 S
OCOR	6.03 qq	_			6.08 qq	5.91 tq	_
	1.98 dq				1.98 dq	5.27 ddq	
	1.90 dq		_		1.91 dq	5.21 ddg	_
	1.504			_	•	1.83 dt	_
OAc	_	_				1.76 s	
ОМе	3.81 s	_					-
ОН	_	-	12.81 s		12.80 s	12.83 s	-

^{* 400} MHz, C₆D₆.

[†] Numbering as in 16.

J (Hz): Compound 16: 3,5 = 2; 5,6 = 8.5; 7,8 = 6.5; 9,10 = 7.5; compounds 17/18:

^{7.8 = 17}; 7.8' = 10.5; compound **20**: 3',4' = 5; 3',5' = 1.5; $4'_1,4'_2 = 12$.

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addition to the coupling J_{CD} that this proton was further coupled with a proton, which after addition of Eu(fod)₃ could be assigned clearly (H_E). The latter was further coupled with the olefinic proton (HA). Therefore the presence of a six-membered ring was established. Further decoupling experiments showed that the iso-propyl group was located α to H_E . As H_D showed no additional coupling, an allylic position was very likely, in agreement with its chemical shift. This, however, led to the structure of a cadinene derivative. The observed couplings $J_{5,6}$ and $J_{5,10}$ required the assigned stereochemistry. As some signals, even at 400 MHz, were still overlapped, compound 23b was degraded. Epoxidation gave, as expected, only one epoxide. Acid-catalysed hydrolysis led to the diol 25 and the elimination product 26, its ¹H NMR spectrum and spin decoupling further supporting the proposed structure and stereochemistry of 23b. The cis position of H-6 and the C-7 hydroxy group in the diol 25 followed from the direction of the elimination of water. As the signals of H-9 were still overlapped, the H-10 signal was not first order. However, spin decoupling established the proposed stereochemistry at C-5, C-6 and C-10. The diol 25 was further degraded to the corresponding ketone 27, but its ¹H NMR spectrum was not very instructive as most signals were overlapped. Furthermore, compound 26 was transformed to the aldehyde 28, which had ¹H NMR data in agreement with the proposed structure.

The aerial parts of *T. scottmorii* afforded germacrene D, γ-selinene (29) and the corresponding ketone 30, the 4-hydroxygermacra-1(10),5-diene (31), the chromene 32, taraxasteryl acetate and again a furanoheliangolide, the acetoxy angelate 3c [11].

The aerial parts of T. salviaefolia afforded germacrene D. taraxastervl acetate and a complex mixture of sesquiterpene lactones, all being of a new type. After repeated TLC four pairs of inseparable, epimeric hemiacetals were obtained, 36a and 36b, 37a and 37b, 38a and 38b and 39a and 39b. The main constituents were 37a and 37b, which on oxidation afforded a single compound, the dilactone 37c. The ¹H NMR data of 37a/b (Table 6) showed the presence of an acetate and an epoxyangelate residue, which must be placed at C-8 and C-9, as irradiation of the H-1 signal caused a sharpening of a doublet which was coupled with a second one. The H-13 signals were singlets, indicating a substituent at C-7. The ¹³C NMR spectrum of the epimers indicated that only one quarternary carbon not bearing an oxygen function was present, while two downfield doublets at 105.4 and 108.7 respectively, indicated hemiacetal carbons. Inspection of models showed that the two epimers

Table 5. ¹H NMR spectral data of compounds 23b, 24 and 26-28 (400 MHz, TMS as internal standard)

		23b						
	CDCl ₃	C ₆ D ₆	+ Eu(fod) ₃	24	26 (C ₆ D ₆)	Δ*(26)	27 (CDCl ₃)	28 (CDCl ₃)
Η-1α	2.14 m	1.50 m	1.58 m		1.73 m	0.07		2.01 br.d
Η-1β	1.95 dddd	1.82 dddd	1.97 dddd		1.42 dddd	0.09		1.40 dddd
Η-2α	2.2 m	2.34 m	2.87 <i>br.ddd</i>		2.18 br.ddd	0.20		2.36 m
Η-2β	2.5 br.dd	2.62 br.dd	3.25 br.dd		2.45 br.ddd	0.13		2.39 br.d
H-4	7.12 ddd	7.31 br.d	7.80 br.d	7.13 br.d	7.19 ddd	0.11	6.91 ddd	7.07 br.d
H-5	2.2 m	2.12 m	2.29 ddd		2.12 ddd	0.13		2.26 ddd
H-6	2.39 br.ddd	2.34 m	2.49 ddd		2.26 br.ddd	0.23	2.77 ddd	2.34 m
H-8	2.20 m	2.12 m	2.12 m		5.53 br.dd	0.30		6.83 br.dd
H-9α H-9β	1.5 4 m 1.73 dddd	1.40 m 1.54 m	\{ 1.55 m	}	1.73 m	0.18		2.36 ddd 2.16 br.dd
H-10	1.10 dddd	1.03 m	1.05 m		1.32 m	0.19		1.67 dddd
H-11	2.0 m	1.95 dqq	2.10 m		1.89 dqq	0.06	$2.04 \ dqq$	2.01 dqq
H-12	0.93 d	0.84 d	0.88 d	0.94 d	0.77 d	0.04	1.04 d	0.96 d
H-13	0.86 d	0.73 d	0.78 d	0.91 d	0.72 d	0.05	0.97 d	0.93 d
H-14	4.70 br.s	4.75 br.s	4.79 br.s	2.64 d	3.91 br.d	0.55		0.42 -
H-14′	4.65 dd	4.73 br.s	4.76 br.s	2.63 d	3.80 br.d	0.55	_	9.42 s
ОМе	3.74 s	3.55 s	4.20 s	3.74 s	3.45 s	0.18	3.74 s	3.75 s

^{*} Δ-values after addition of Eu(fod)₃.

obviously differed in the stereochemistry at the hemiacetal carbon. In the case of the α-orientated hydroxyl (37a) H-1 was deshielded, while the β -hydroxy group (37b) deshielded the 4-methyl group. A smaller effect on the chemical shifts of H-8, H-9 and H-13 was observed. In the spectrum of the dilactone 37c the expected shifts were obtained (Table 6). In particular, the downfield shift of the C-4 methyl signal supported the position of the new lactone carbonyl. The stereochemistry at C-6 was deduced from the chemical shift of H-6, which obviously was not much affected by the orientation of the 5hydroxyl group, while a 6β -proton should be deshielded strongly by a 5β -hydroxy group. The observed Eu(fod)₃induced shifts further supported the proposed stereochemistry. The ¹H NMR spectra of the three other pairs of epimers were similar to those of 37a/b. The spectrum of the epimers 36a/b showed that the epoxyangelate was replaced by an angelate residue. Compounds 38a/b and 39a/b obviously had a free 9hydroxy group, as the broadened doublet (allylic coupling with H-1) was shifted to higher fields. This was the main argument for the relative position of the ester groups in 36a/b and 37a/b. A small shift difference for H-8 in the spectra of these two pairs further supported this assumption. We propose the name trichosalviolide for the parent compound of these lactones without oxygen functions at C-5, C-8 and C-9 (numbering as in a germacranolide). Most probably these lactones are formed as shown in Scheme 4 starting with the epoxide 33 which was formed by oxidation of costunolide. A Wagner-Meerwein rearrangement of the opened epoxide and subsequent attack of the aldehyde carbonyl at C-7, transformed to a cation by protonation of the hydroxyl, would lead to the natural compounds.

The roots afforded germacrene D, bicyclogermacrene and again several p-hydroxyacetophenone derivatives, the vinyl compounds 16, 17 and 18a as well as the phenols 19-21. In addition to these compounds, the methyl ether of 18a (18b) and the enol ether of 17 (40) were isolated. The structures followed from the mass spectra and the ¹H NMR data. In the ¹H NMR spectrum of 18b some signals were shifted slightly, compared with those of 18a (Table 4), while the ¹H NMR data of 40 (see Experimental) showed that no aromatic ketone was present. Consequently the signals of the aromatic protons were at higher fields. The position of the methoxy group followed from the chemical shift of the olefinic proton.

The roots of *T. villosa* afforded germacrene D, **18a**, **19**, **20** and **40**, while the aerial parts gave germacrene D, bicyclogermacrene, α-humulene, aromadendrene, taraxasteryl acetate, **18a** and the furanoheliangolides **1b** and **41**. The ¹H NMR data of **1b** were similar to those of **1a** (Table 1) except for the signals of the different ester residues. The ¹H NMR spectrum of **41** showed some similarity to that of **1b** (Table 1). The presence of a methyl doublet, however, and the signals of H-4 through H-6 and of H-15

J (Hz): Compound 23b: $1\alpha,1\beta=13$; $1\alpha,2\alpha=6$; $1\alpha,6=12$; $1\beta,2\alpha=12$; $1\beta,2\beta=6$; $1\beta,6=12$; $2\alpha,4=2\beta,4\sim1.5$; 4,5=5; 5,6=6; 5,10=12; $8\alpha,9\beta=8\beta,9\beta=3.5$; $8,14'\sim1.5$; $9\alpha,9\beta=13$; $9\alpha,10=12$; $9\beta,10=4$; 10,11=12; 11,12=11,13=7; compound 26: $1\alpha,1\beta=10$; $1\alpha,2\alpha=1\alpha,2\beta=4$; $1\beta,2\alpha=10$; $1\beta,2\beta=5$; $1\beta,6=10$; $2\alpha,2\beta=17$; $2\alpha,4=2\beta,4\sim1.5$; 4,5=5; $5,6\sim5$; 5,10=10; 8,9=3.5; $8,14'\sim1.5$; $10,11\sim10$; 11,12=11,13=7; compound 28: $1\alpha,1\beta=12$; $1\beta,2\alpha=1\beta,6=12$; $1\beta,2\beta=5$; 4,5=5; 5,6=6; 5,10=10; $8,9\alpha=5$; $8,9\beta=1.5$; $9\alpha,9\beta=19$; $9\alpha,10=10$; $9\beta,10=5$; 10,11=10; 11,12=11,13=7.

Table 6. ¹H NMR spectral data of compounds 36a/b-39a/b

		CDCl ₃	, 400 MHz								
	36a	36b	37a	37b	Δ	۸*	37c	38a	38b	39a	39b
H-1	6.01 br.da	l 5.68 br.da	l 6.03 br.da	l 5.68 br.dd	0.29	0.21	5.52 br.dd	5.84 br.d	5.55 br.d	5.86 br.d	5.55 br.d
H-2	2.3	m	2.45	m							
H-2'	2.25	ddd	2.34	br.d			2.53 m	2.4-2	2.1 m	2.4-2	2.1 m
H-3	1.65	br.dd	1.66	br.dd			2.42 m	1.68	br.dd	1.	68 m
H-3'	2.3	m	2.25	ddd			2.05 m				
H-5	4.93 s	5.39 s	4.93 d	5.39 d	0.23	0.25	_	4.93 s	5.38 s	4.92 d	5.38 d
H-6	4.92 s	4.87 s	4.92 s	4.88 s	0.35	0.35	5.11 s	4.93 s	4.88 s	4.92 s	4.88 s
H-8	5.69 d	5.73 d	5.66 d	5.71 d	0.77	0.77	5.73 d	5.60 d	5.63 d	5.63 d	5.67 d
H-9	4.79 br.d	6.75 br.d	6.80 br.d	5.77 br.d	0.78	0.76	5.95 d	4.85 d	4.80 br.d	4.83 d	4.79 d
H-13	6.45 s	6.40 s	6.42 s	6.37 s	0.29	0.31	6.61 s	6.46 s	6.40 s	6.43 s	6.37 s
H-13'	$6.12 \ s$	6.07 s	6.11 s	6.05 s	0.27	0.31	6.28 s	6.09 s	6.03 s	6.08 s	$6.02 \ s$
H-14	1.79 br.s	1.82 br.s	1.82 br.s	1.85 br.s	0.26	0.27	1.89 br.s	1.78 br.s	1.82 br.s	1.81 br.s	1.84 br.s
H-15	0.96 s	2.02 s	0.97 s	1.03 s	0.14	0.22	1.11 s	0.97 s	1.03 s	0.97 s	1.03 s
OAc	2.02 s	2.03 s	2.04 s	2.05 s	0.26	0.34	2.05 s		11 Miles	-> Note:	
OCOR	6.0	7 gg	3.0	0 <i>q</i> †	0.	46	$3.03 \; q$	6.14 qq		3.0	0 q
	1.9	2 dq	1.19 d		•		1.97 dq			9 d	
		2 dq	1.9	5 s	0.	32	1.53 s	1.82 <i>br.s</i>		1.5	3 s
ОН			3.65 d	2.66 d			_				

^{*} Δ -values after addition of Eu(fod)₃.

R

R'

R"

Ac

Ang

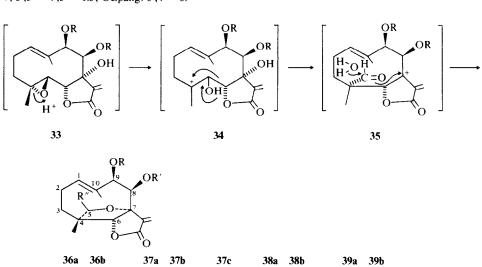
α-ΟΗ β-ΟΗ

Ac

Epang

 α -OH β -OH

J (Hz): 1,2 = 12; 1,2' = 3.5; 1,9 ~ 0.5; 2,3 ~ 2; 2',3 = 8; 3,3' = 14; 5,OH = 4.5 (compound 37b: 2.5); 8,9 = 4; OAng: 3',5' = 7; 3',5' = 4',5' = 1.5; OEpang: 3',4' = 5.



Scheme 4.

Н

Ang

α-ΟΗ β-ΟΗ

Н

Epang

α-ΟΗ β-ΟΗ

Ac

Epang

=0

[†] In C₆D₆: 2.66 and 2.64 q, 1.24 and 1.22 d, 1.47 and 1.45 s.

were typically changed indicating a hydrogenated 4,5-double bond. The stereochemistry at C-4 was probably the same as in zexbrevin [12], an 8-O-methacrylate.

The overall picture of the chemistry of the genus *Trichogonia* showed that *p*-hydroxyacetophenone derivatives are characteristic, especially those where the aceto group is transformed to a vinyl group. Furthermore, the occurrence of highly oxidized germacranolides and their transformation products seems to be typical. Furanoheliangolides have been isolated from this group so far only from *Conocliniopsis* [13]. Highly oxygenated sesquiterpene lactones are also present in *Lasiolaena* [14] and *Agrianthus* [15], while *p*-hydroxyacetophenone derivatives were isolated from a *Bahianthus* [16] and a *Campuloclinium* species (F. Bohlmann *et al.*, unpublished results). This taxonomically somewhat diverse group therefore seems also not to be very uniform chemically. However, so far not many genera have been investigated.

EXPERIMENTAL

The air-dried plant material was extracted with $\rm Et_2O$ -petrol (1:2) and the resulting extracts were first separated by CC (Si gel) and further by repeated TLC (Si gel). Known compounds were identified by comparing the IR and 1H NMR spectra with those of authentic material.

Trichogonia prancii Barroso (voucher RMK 8274). The aerial parts (430 g) afforded 5 mg germacrene D, 5 mg bicyclogermacrene, 5 mg α-humulene, 100 mg taraxasterol and 80 mg taraxasteryl acetate, 30 mg 1a (Et₂O-petrol, 1:3), 2 mg 2 (Et₂O-petrol, 3:1), 8 mg 3a (Et₂O-petrol, 3:1), 25 mg 4 (Et₂O-petrol, 3:1), 5 mg 5 (Et₂O-petrol, 3:1), 5 mg 6 (Et₂O-petrol, 3:1), 15 mg 7 (Et₂O-petrol, 3:1) and 5 mg 8a (Et₂O-petrol, 3:1). (1-8a were further purified by TLC using CHCl₃-Et₂O, 1:1.) The roots (260 g) yielded 4 mg α-humulene, 7 mg bicyclogermacrene, 10 mg lupeyl acetate, 20 mg 9 (Et₂O-petrol, 1:3) and 15 mg 10a (Et₂O-petrol, 1:3).

Trichogonia grazielae K. et R. (voucher RMK 8271). The roots (150 g) afforded 2 mg germacrene D, 5 mg 16 (Et₂O-petrol, 1:10), 2 mg 17 (Et₂O-petrol, 1:10), 15 mg 18a (Et₂O-petrol, 1:3), 15 mg 19 (Et₂O-petrol, 1:3), 60 mg 20 (Et₂O-petrol, 1:3), 2 mg 21 and 1 mg 22 (Et₂O-petrol, 3:1), while the aerial parts (200 g) gave 10 mg germacrene D, 10 mg α -humulene, 30 mg aromadendrene, 20 mg caryophyllene, 100 mg herniarin, 50 mg 13 and 400 mg 23a (Et₂O-petrol, 1:3).

Trichogonia scottmorii K. et R. (voucher RMK 8018). The aerial parts (50 g) afforded 10 mg germacrene D, 10 mg taraxasteryl acetate, 50 mg 3c, 10 mg 29, 15 mg 30, 10 mg 31 and 5 mg 32.

Trichogonia salviaefolia Gardn. (voucher RMK 8183). The aerial parts (300 g) afforded 30 mg germacrene D, 50 mg taraxasteryl acetate, 1 g fatty acids, 15 mg 36a/b, 100 mg 37a/b, 5 mg 38a/b and 10 mg 39a/b (all purified by TLC, Et₂O-petrol, 3:1 and then CHCl₃-Et₂O, 1:1), while the roots (80 g) gave 10 mg germacrene D, 2 mg bicyclogermacrene, 2 mg 16, 3 mg 17,

 $30 \text{ mg } 18a, 2 \text{ mg } 18b \text{ (Et}_2O-\text{petrol}, 1:3), 8 \text{ mg } 19,60 \text{ mg } 20,10 \text{ mg } 21 \text{ and } 2 \text{ mg } 40 \text{ (Et}_2O-\text{petrol}, 1:10).}$

Trichogonia villosa (DC) Sch. Bip. ex Baker (voucher RMK 8361). Roots (29 g) afforded 5 mg germacrene D, 30 mg 18a, 5 mg 19, 40 mg 20 and 3 mg 40, while the aerial parts (60 g) gave 40 mg germacrene D, 5 mg bicyclogermacrene, $10 \text{ mg } \alpha$ -humulene, 3 mg aromadendrene, 50 mg taraxasteryl acetate, 5 mg 1b (Et₂O-petrol, 3:1), 5 mg 18a and 5 mg 41 (Et₂O-petrol, 3:1).

9β-Hydroxyatripliciolide-8-O-methacrylate (1a). Colourless gum, not separated from 4. IR $v_{\rm med}^{\rm CCl_4}$ cm⁻¹: 3540 (hydrogenbonded OH), 1783 (γ-lactone), 1730, 1650 (C=CCO₂R), 1705, 1595 (C=C-C=C-OR); MS m/z (rel. int.): 360. 121 (M⁺, 11) (C₁₉H₂₀O₇), 342 (M - H₂O, 1), 274 (M - RCO₂H, 3), 69 (C₃H₅CO⁺, 100). To 10 mg of a mixture of 1a and 4 in 1 ml CHCl₃, 10 mg 4-pyrrolidinopyridine and 0.1 ml Ac₂O were added. After 12 hr, TLC (Et₂O-petrol, 3:1) afforded 10 mg of 2 and 5, which could not be separated. Both compounds were identical (¹H NMR spectra) with the natural compounds.

9 β -Hydroxyatripliciolide-8-O-angelate (1b). Colourless crystals, mp 75° (petrol); IR $\nu_{max}^{\text{CCl}_{1}}$ cm $^{-1}$: 3530 (OH), 1785 (γ -lactone), 1735, 1655 (C=CCO $_2$ R), 1710, 1600 (O=C-C=C-OR); MS m/z (rel. int.): 374.137 (M $^+$, 10) (C $_2$ 0 H_2 2O $_7$), 356 (M-H $_2$ O, 0.5), 274 (M-RCO $_2$ H, 3), 83 (C $_4$ H $_7$ CO $^+$, 100), 55 (83 - CO, 74).

9β-Acetoxyatripliciolide-8-O-methacrylate (2). Colourless gum, IR $v_{\rm max}^{\rm CCl_4}$ cm⁻¹: 1770 (γ-lactone), 1740 (OAc), 1715 (C=CCO₂R, C=O), 1595 (C=COR); MS m/z (rel. int.): 402. 131 (M⁺, 16) (C₂₁H₂₂O₈), 360 (M – ketene, 5), 343 (M – OAc, 2), 274 (360 – RCO₂H, 5), 232 (274 – C₂H₂O, 18), 69 (C₃H₅CO⁺, 100).

9β-Methacryloyloxyatripliciolide (3a). Colourless crystals, mp 164°. IR $v_{max}^{\rm CCI_4}$ cm $^{-1}$: 3580 (OH), 1785 (γ-lactone), 1720, 1650 (C=CCO₂R), 1720, 1605 (O=CC=COR); MS m/z (rel. int.): 360.121 (M $^+$, 12) (C₁₉H₂₀O₇), 274 (M $^-$ RCO₂H, 1), 256 (274 $^-$ H₂O, 1), 232 (274 $^-$ C₂H₂O, 17), 69 (C₃H₅CO $^+$, 100).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{-82.5} \frac{578}{-87.5} \frac{546}{-101.0} \frac{436 \text{ nm}}{-199.5} (c = 0.2, \text{CHCl}_3).$$

Compound 3a (5 mg) was acetylated as above. TLC (Et₂O-petrol, 3:1) afforded 5 mg 3b, colourless gum. IR $\nu_{\rm max}^{\rm CCI_4}$ cm⁻¹: 1785 (γ -lactone), 1730 (OAc, C=CCO₂R, C=O), 1610 (C=COR); MS m/z (rel. int.): 402.131 (M⁺, 35) (C₂₁H₂₂O₈), 360 (M - ketene, 6), 342 (M - HOAc, 2), 300 (342 - C₂H₂O, 9), 274 (360 - RCO₂H, 4), 232 (274 - C₂H₂O, 19), 69 (C₃H₅CO⁺, 100).

9β-Hydroxy-11β,13-epoxyatripliciolide-8-O-methacrylate (4). Colourless gum, not separated from 1. IR $\nu_{\rm max}^{\rm CCI_4}$ cm⁻¹: 3540 (hydrogen-bonded OH), 1805 (γ-lactone), 1730, 1650 (C=CCO₂R), 1705, 1595 (O=C-C=C-OR); MS m/z (rel. int.): 376.116 (M⁺, 5) (C₁₉H₂₀O₈), 290 (M – RCO₂H, 1), 69 (C₃H₅CO⁺, 100).

 9β -Acetoxy-11 β ,13-epoxyatripliciolide-8-O-methacrylate (5). Colourless gum, not separated from 2. IR $v_{\rm max}^{\rm CCl4}$ cm $^{-1}$: 1795 (γ-lactone), 1740 (OAc), 1715 (C=CCO $_2$ R, C=O), 1590 (C=COR); MS m/z (rel. int.): 418.126 (M $^+$, 12) (C $_2$ 1 $_1$ 2 $_2$ 0 $_9$), 376

 $(M - \text{ketene}, 6), 291 (376 - RCO_2, 6), 69 (C_3H_5CO^+, 100).$

Trichogoniolide (6). Colourless gum. IR $v_{\text{max}}^{\text{CHC}_3}$ cm⁻¹: 3620 (OH), 1765 (γ-lactone), 1720 (C=CCO₂R, lactone), 1645, 1610 (C=C); MS m/z (rel. int.): 380.147 (M⁺, 4) (C₁₉H₂₄O₈), 362 (M - H₂O, 1), 336 (M - CO₂, 1), 69 (C₃H₅CO⁺, 100). Compound 6 (5 mg) was acetylated as above. TLC (Et₂O-petrol, 3:1) afforded 4 mg 7, identical with the natural acetate.

Trichogoniolide-9-O-acetate (7). Colourless crystals, mp 215° (Et₂O). IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 3600, 3420 (OH), 1778 (γ-lactone), 1755(OAc), 1725 (C=CCO₂R, lactone), 1640, 1605 (C=C); MS m/z (rel. int.): 422.158 (M⁺, 0.5) (C₂₁H₂₆O₉), 404.147 (M - H₂O, 1) (C₂₁H₂₄O₈), 380 (M - ketene, 0.5), 362 (M - HOAc, 1), 347 (362 - Me, 0.5), 336 (M - RCO₂H, 3), 294 (336 - ketene, 2), 276 (336 - HOAc, 4), 258 (276 - H₂O, 2), 69 (C₃H₅CO⁺, 100).

$$[\alpha]_{24}^{\lambda} = \frac{589}{-246.2} \frac{578}{-256.5} \frac{546}{-293.9} \frac{436}{-520.1} \frac{365 \text{ nm}}{-857.2}$$

$$(c = 0.69, \text{CHCl}_3).$$

Isotrichogoniolide-9-O-acetate (8a). Colourless gum. IR $\nu_{\rm max}^{\rm CCl_*}$ cm⁻¹: 3620 (OH), 1780 (γ-lactone), 1735 (C=CCO₂R, lactone), 1640 (C=C); MS m/z (rel. int.): 422.158 (M⁺, 0.5) (C₂₁H₂₆O₉), 404.147 (M − H₂O, 1) (C₂₁H₂₄O₈), 380 (M − kctene, 1), 362 (M − HOAc, 1), 344 (404 − HOAc, 0.5), 336 (M − RCO₂H, 4), 318 (336 − H₂O, 4), 276 (336 − HOAc, 4), 241 (276 − MeCHOH, 6), 232 (276 − CO₂, 7), 69 (C₃H₅CO⁺, 100). Compound 8a (5 mg) was acetylated as above. TLC(Et₂O) afforded 4 mg 8b, colourless gum. IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 1770 (γ-lactone), 1745 (OAc), 1720 (C=CCO₂R, lactone); MS m/z (rel. int.): 464 (M⁺, 0.3), 404.147 (M − HOAc, 8) (C₂₁H₂₆O₉), 318 (404 − RCO₂H, 2), 69 (C₃H₅CO⁺, 100).

2-[4'-Formylfuryl]-4-vinylanisole (9). Colourless crystals, mp 159° (petrol). IR $v_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 2725, 1690 (furan aldehyde), 1630, 910 (CH=CH₂); MS m/z (rel. int.): 228.078 (M $^+$, 72) (C₁₄H₁₂O₃), 213 (M – Me, 6), 185.060 (213 – CO, 33) (C₁₂H₉O₂), 157.065 (185 – CO, 100) (C₁₁H₉O), 128.063 (157 – CHO, 97) (C₁₀H₈). To 5 mg 9 in 1 ml MeOH, 10 mg NaBH₄ was added. TLC (Et₂O-petrol, 1:1) afforded 3 mg 10b, colourless gum. IR $v_{\rm max}^{\rm CCl_5}$ cm $^{-1}$: 3600 (OH), 1635, 920 (CH=CH₂); MS m/z (rel. int.): 230 (M $^+$, 100), 215 (M – Me, 2), 201 (M – CHO, 3), 187 (215 – CO, 10).

2-[4'-Acetoxymethylfuryl]-4-vinylanisole (10a). Colourless gum, IR $v_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 1745, 1260 (OAc), 1630, 910 (CH=CH $_2$); MS m/z (rel. int.): 272.155 (M $^+$, 100) (C $_{16}$ H $_{16}$ O $_4$), 230 (M $_2$ ketene, 16), 215 (230 - Me, 9), 201 (230 - CHO, 33), 187 (215 - CO, 18), 128 (C $_{10}$ H $_8$, 11). Compound 10a (8 mg) in 3 ml Et $_2$ O was hydrogenated in the presence of palladium on BaSO $_4$ (1 hr). TLC (Et $_2$ O-petrol, 1:1) afforded 6 mg 12a, colourless gum. IR $v_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 3620 (OH), 1740, 1250 (OAc); MS m/z (rel. int.): 280 (M $^+$, 28), 162 (EtC $_6$ H $_3$ (OMe)CH=CH $_2$ H $_7$, 70), 149 (EtC $_7$ H $_5$ OMe $_7$, 100), 119 (149 - CH $_2$ O, 45). Compound 12a (6 mg) was acetylated as above, yielding 6 mg 12b. For 1 H NMR data see Table 2.

2-[3',3'-Dimethylallyl]-4-[1-angeloyloxyethyl]-anisole (16). Colourless oil. IR $v_{\rm max}^{\rm CCl_+}$ cm $^{-1}$: 1715, 1655 (C=CCO $_2$ R), 1615, 1505 (aromate); MS m/z (rel. int.): 302.188 (M $^+$, 19) (C $_{19}$ H $_{26}$ O $_{3}$), 203 (M — OAng, 100), 202 (M — AngOH, 28), 187 (202 — Me, 68), 83 (C $_4$ H $_7$ CO $^+$, 28).

2,2-Dimethyl-6-vinylchroman-4-one (17). Colourless oil. IR $v_{\rm max}^{\rm CCla}$ cm $^{-1}$: 1700, 1615 (PhCO), 1635, 915 (CH=CH₂); MS m/z (rel. int.): 202.099 (M⁺, 56) (C₁₃H₁₄O₂), 187 (M - Me, 100), 146 (M - Me₂C=CH₂, 74), 118 (146 - CO, 20).

2-Senecioyl-4-vinylphenol (18a). Colourless oil. IR $v_{\text{max}}^{\text{CCL}_4}$ cm⁻¹: 3500–2700, 1630 (hydrogen-bonded hydroxyketone), 1610, 915 (CH=CH₂); MS m/z (rel. int.): 202.099 (M⁺, 11) (C₁₃H₁₄O₂),

 $187 (M - Me, 100), 147 (M - CH=CMe_2, 15).$

2-Senecioyl-4-vinylanisole (18b). Colourless oil. IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1655 (C=CCOPh), 1620, 1495 (aromate), 910 (CH=CH₂); MS m/z (rel. int.): 216.115 (M⁺, 78) (C₁₄H₁₆O₂), 201 (M - Me, 100), 187 (M - CHO, 91).

2-Senecioyl-4- $\{1\text{-angeloyloxyethyl}\}$ -phenol (19). Colourless oil. IR $v_{\max}^{\text{CCl}_4}$ cm⁻¹: 3500–2700, 1640 (hydrogen-bonded ketone), 1715 (C=CCO₂R); MS m/z (rel. int.): 302.152 (M⁺, 8) (C₁₈H₂₂O₄), 287 (M - Me, 18), 203 (M - OAng, 28), 187 (287 - AngOH, 100), 147 (203 - CH=CMe₂, 52), 83 (C₄H₇CO⁺, 21). 2-Senecioyl-4- $\{1\text{-}(4\text{-acetoxyangeloyloxyethyl-}\}$ -phenol (20). Yellow oil. IR $v_{\max}^{\text{CCl}_4}$ cm⁻¹: 3400–2600, 1645 (hydrogen-bonded ketone), 1747, 1225 (OAc), 1715 (C=CCO₂R); MS m/z (rel. int.): 360 (M⁺, 0.2), 345.134 (M - Me, 6) (C₁₉H₂₁O₆), 203 (M - O₂CR, 55), 187 (345 - RCO₂H, 100), 147 (203 - CH₂= CMe₇, 32).

$$[\alpha]_{24^{\circ}}^{\lambda} = \frac{589}{+3.6} \frac{578}{+4.0} \frac{546 \text{ nm}}{+4.5} (c = 6.0, \text{ CHCl}_3).$$

2,2-Dimethyl-6-[1-hydroxyethyl]-chroman-4-one (22). Colourless oil. IR $v_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 3630 (OH), 1705, 1625 (PhCO); MS m/z (rel. int.): 220.110 (M $^+$, 31) (C $_{13}$ H $_{16}$ O $_3$), 205 (M $^-$ Me, 88), 149 (205 $^-$ CH $_2$ =CMe $_2$, 100).

 γ -Muurolen-15-oic acid (23a). Colourless oil, IR $\nu_{\rm max}^{\rm CCL_4}$ cm $^{-1}$: 3400–2500, 1692, 1645 (C=CCO₂H); MS m/z (rel. int.): 234.162 (M $^+$, 38) (C_{1.5}H_{2.2}O₂). 191 (M - C₃H₇, 100); 13 C NMR (CDCl₃) (C-1 through C-15): 24.7 t, 26.1 t, 129.4 s, 144, 7 d, 40.3 d, 42.3 d, 151.9 s, 31.2 t, 24.7 t, 33.8 d, 27.2 d, 21.4 q, 15.8 q, 107.8 t, 171.8 s (doublets and triplets assigned by partial decoupling). Addition of CH₂N₂ in Et₂O afforded 23b (TLC, Et₂O-petrol, 1:10), colourless oil. IR $\nu_{\rm max}^{\rm CCL_4}$ cm $^{-1}$: 3060, 900 (C=CH₂), 1720, 1645 (C=CCO₂R); MS m/z (rel. int.): 248.197 (M $^+$, 82) (C_{1.6}H_{2.4}O₂), 233 (M - Me, 5), 217 (M - OMe, 11) 205 (M - C₃H₇, 100), 189 (217 - CO, 22), 173 (205 - MeOH, 71), 145 (173 - CO, 70).

$$[\alpha]_{24^{\circ}}^{2} = \frac{589}{+28.1} \frac{578}{+19.2} \frac{546}{-22.9} \frac{436 \text{ nm}}{+50.0} (c = 8.2, \text{ CHCl}_3).$$

To 50 mg 23b in 2 ml CHCl₃, 75 mg *m*-chlorperbenzoic acid was added. After 2 hr the reaction product (24), was isolated, colourless oil. Its ¹H NMR spectrum (Table 5) indicated that only one epoxide had been formed.

The crude epoxide (24) was warmed in MeOH-2NH₂SO₄ (2:1) for 30 min at 70°, affording 15 mg 26 (TLC: Et₂O-petrol, 1:1) and 25 mg 25, which was directly reacted with NaIO₄. TLC (Et₂O-petrol, 1:3) gave 12 mg 27, colourless oil. IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 1720 (C=CCO₂R, C=O); MS m/z (rel. int.): 250.157 (M⁺, 100) $(C_{15}H_{22}O_3)$, 219 (M - OMe, 60), 218 (M - MeOH, 91), 175 $(218 - C_3H_7, 95)$. **26**: Colourless oil. IR $v_{\text{max}}^{\text{CCl}_4}$ cm⁻¹: 3600 (OH), 1720 (C=CCO₂R); MS m/z (rel. int.): 264.172 (M⁺, 6) $(C_{16}H_{24}O_3)$, 246 $(M - H_2O, 55)$, 233 (M - OMe, 32), 203 (246) $-C_3H_7$, 48), 134 (100). Compound **26** (10 mg) was stirred with 100 mg MnO₂ for 1 hr. TLC (Et₂O-petrol, 1:3) afforded 8 mg 28, colourless oil. IR $v_{\text{max}}^{\text{CCI}_4}$ cm⁻¹: 2720, 1690 (CHO), 1720, 1645 $(C=CCO_2R); MSm/z (rel. int.): 262.157 (M^+, 49), 244 (M - H_2O_1)$ 16), 230 (M - MeOH, 66), 219 (M - C_3H_7 , 27), 202 (230 - CO, 15), $187(230 - C_3H_7, 100)$, 159(187 - CO, 63), 131(159 - CO, 63)50), 91 (C₇H₇⁺, 41).

9 β -Acetoxy-8 β -angeloyloxy-5 α - and 5 β -hydroxytrichosalviolide (36a/b). Colourless crystalline mixture, mp 144° (Et₂O-petrol). IR $\nu_{\rm max}^{\rm CCl_4}$ cm⁻¹: 3600 (OH), 1773 (γ -lactone), 1750, 1237 (OAc), 1730 (CO₂R)m, 1660 (C=CH₂); MS m/z (rel. int.): 420.178 (M⁺, 0.2) (C₂₂H₂₈O₈), 378 (M - ketene, 1), 360 (M - HOAc, 4), 278 (378 - AngOH, 6), 83 (C₄H₇CO⁺, 100), 55 (83 - CO, 42); CI (isobutane): 421 (M + 1,8), 403 (421 - H₂O, 13), 361 (421 - HOAc,

100), 303 (403 - AngOH, 58), 285 (303 - H_2O , 23), 261 (361 - AngOH, 20), 243 (261 - H_2O , 17), 83 ($C_4H_7CO^+$, 59).

 $9B-Acetoxy-8B-[2-methyl-2.3-epoxybutyryloxy]-5\alpha$ and 5Bhydroxytrichosalviolide (37a/b). Colourless crystalline mixture, mp 224° (Et₂O). IR $v_{max}^{CHCl_3}$ cm⁻¹: 3600 (OH), 1770 (y-lactone), 1730 (OAc, CO₂R); MS m/z (rel. int.): 436.173 (M⁺, 0.3) $(C_{22}H_{28}O_9)$, 418 $(M - H_2O, 1)$, 394 (M - ketene, 3), 376 (M- HOAc, 6), 358 (376 - H₂O, 1), 320 (M - RCO₂H, 4), 302 (320 - H₂O, 5), 278 (320 - ketene, 36), 260 (320 - HOAc, 37), 232 (260 - CO, 70), 99 (RCO⁺, 30), 71 (99 - CO, 100); ¹³C NMR (CDCl₂) (C-1 through C-15): 129.6 (129, 3), 30.9, 38.5, 47.9, (50.7), 105.4 (108.7), 84.2 (86.1), 87.7 (83.6), 76.4 (77.3), 76.5 (76.1), 130.7, 138.9 (140.2), 168.4, 128.0 (127.6), 19.1, 13.6 (13.2) (some signals may be interchangeable). Compounds 37a/b (5 mg) in 2ml CHCl₃ were stirred for 2 hr with 50 mg pyridine dichromate. TLC (Et₂O-CHCl₃, 1:2) afforded 3 mg 37c, colourless crystals, mp 212° (Et₂O-petrol). IR $v_{max}^{CCl_4}$ cm⁻¹: 1790, 1783 (y-lactone), 1755 (OAc), 1730 (CO₂R); MS m/z (rel. int.): 434.158 (M⁺, 6) (C₂₂H₂₆O₉), 392 (M - ketene, 10), 374 (M - HOAc, 6), $276 (392 - RCO_2H, 100)$, $258 (276 - H_2O, 20)$, 248 (276 - CO, 21), 230 (258 - CO, 26), 99 (RCO+, 20), 71 (99

$$[\alpha]_{24}^{\lambda}$$
 = $\frac{589}{-5.6}$ $\frac{578}{-8.9}$ $\frac{546}{-8.9}$ $\frac{436 \text{ nm}}{-11.1}$ ($c = 0.1$, CHCl₃).

8β-Angeloyloxy-5α- and 5β,9β-dihydroxytrichosalviolide (38a/b). Colourless gum. IR $v_{\rm max}^{\rm COl_4}$ cm⁻¹: 3600 (OH), 1775 (γ-lactone), 1730 (CO₂R); MS m/z (rel. int.): 378 (M⁺, 0.1), 360.157 (M - H₂O, 0.5) (C₂₀H₂₄O₆), 278 (M - AngOH, 7), 83 (C₄H₇CO⁺, 100), 55 (83 - CO, 54).

8β-[2-Methyl-2,3-epoxybutyryloxy]-5α- and 5β,9β-dihydroxytrichosalviolide (39a/b). Colourless gum. IR $v_{\rm max}^{\rm CHCl_3}$ cm⁻¹: 1665, 1620 (C=CCOPh), 910 (CH=CH₂); MS m/z (rel. int.): 376.152 (M – H₂O, 1), 278 (M – RCO₂H), 99 (RCO⁺, 11), 71 (99 – H₂O, 100).

4-Methoxy-6-vinyl-2,3-dimethylchromene (40). Colourless oil. IR $v_{\rm max}^{\rm CCl+}$ cm⁻¹: 3070, 1630, 905 (CH=CH₂), 1650 (C=COR), 1600, 1490 (aromate); MS m/z (rel. int.): 216.115 (M⁺, 14) (C₁₄H₁₆O₂), 186 (M - CH₂O, 7), 171 (186 - Me, 2); ¹H NMR (CDCl₃): 1.45 (s, H-12 and 13), 3.70 (s, OMe), 6.64 (dd, J = 17.5, 11 Hz, H-7), 5.62 (dd, J = 17.5, 1 Hz, H-8t), 5.11 (dd, J = 11, 1 Hz, H-8c), 7.47 (d, J = 2 Hz, H-3), 7.22 (dd, J = 8 Hz, H-5), 6.75 (d, J = 8 Hz, H-6), 4.64 (s, H-10).

8 β -Angeloyloxy-9 β -hydroxyzexbrevanolide (41). Colourless gum. IR $\nu_{\rm max}^{\rm CCl_4}$ cm $^{-1}$: 3600 (OH), 1780 (γ -lactone), 1730 (C=CCO₂R) 1700 (C=CCO), 1600 (C=COR); MS m/z (rel. int.): 376.152 (M $^+$, 4) (C₂₀H₂₄O₇), 358 (M - H₂O, 0.5), 276 (M - AngOH, 0.5), 83 (C₄H₇CO $^+$, 100).

X-ray analysis of 7. The compound crystallizes in the orthorhombic system, space group P2₁2₁2₁, with lattice

constants a=21.214 (5), b=12,110 (3) and c=8.594 (2) Å, Z=4. Single crystal X-ray data were taken with a Syntex P2₁ diffractometer (monochromated MoK_a radiation), using a variable-speed ω -scan technique. 1538 unique reflections with $2\theta \le 45^\circ$ were measured, of which 1400 with intensities $I>2\sigma_I$ were considered as observed. The structure was solved by direct methods (MULTAN). Full matrix isotropic refinement of all non-hydrogen atoms gave an R-value of 10%. Refinement is still in progress. Details of the X-ray structure determination will be published elsewhere. Fig. 1 shows a perspective drawing of the compound. For reasons of clarity, only the essential carbon atoms are labelled. (The radii of the spheres do not represent isotropic temperature factors, but are different according to perspective height.)

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